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An efficient and convenient protocol for the synthesis of tetracyclic isoindolo[1,2-*a*]quinazoline derivatives†

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A convenient and one-pot synthesis of tetracyclic isoindolo [1,2-*a*]quinazoline derivatives *via* Lewis acid mediated sequential C–N bond formation reactions is reported. This protocol provides a simple and rapid strategy for the synthesis of 12-benzylidene-10,12-dihydroisoindolo[1,2-*b*]quinazoline derivatives. However, a variety of tetracyclo indole fused quinazoline motifs were synthesized in good yields.

Isoindoloquinazolinones symbolize the core structure in numerous biologically active molecules.¹ In addition, they are also important building blocks of potential drug molecules and natural products such as camptothecin **1**, belotecan (CKD-602) **2**,² batracylin **3**,³ tryptanthrin **4**,⁴ ophiuroidine **5**,⁵ (–)-vasicine **6**,⁶ luotonin **7a**, **7b** & **7c**,⁷ and auranthine **8** (Fig. 1).⁸ Isoindoloquinazolinones have been reported with anti-cancer, anti-viral, anti-tubercular and anti-malarial activities. Recently Yang and co-workers⁹ reported that substituted quinazolines have novel potent and selective FLT3 inhibitory and anti-acute myeloid leukaemia (AML) activities.

Because of varied biological properties of quinazolinone derivatives, it is necessary to develop efficient and convenient methods to prepare isoindoloquinazolinone derivatives. Throughout the course of our literature survey we found minimum number of reports for the preparation of isoindoloquinazoline derivatives. Mitscher *et al.* have described intramolecular Aza-Wittig reaction using triethylamine,¹⁰ Weaver *et al.* have reported oxidative radical cyclization for synthesis of quinazolines from quinazolin-4(3*H*)-one.¹¹

The development of simple methodology for the preparation of isoindoloquinazolinone derivatives is always in demand. In the past, our group described numerous protocols for the preparation of quinazolinone based natural products and their derivatives.¹²

In this communication, we wish to report simple and straight forward synthesis of poly-substituted isoindoloquinazolinones derivatives.

The synthetic strategy employed for the synthesis of (*Z*)-12-benzylidene-10,12-dihydroisoindolo[1,2-*b*]quinazoline derivatives is depicted in Scheme 1. The (*Z*)-12-benzylidene-10,12-dihydroisoindolo[1,2-*b*]quinazolines derivatives **11a** could be easily obtained by a reaction of (2-aminophenyl)methanol **9a**¹³ with 2-(phenylethynyl) benzonitrile **10a**.¹⁴

The compound **11** was characterized by ¹H NMR, ¹³C NMR, HRMS and IR. Substituted (*Z*)-12-benzylidene-10,12-dihydroisoindolo [1,2-*b*]quinazoline derivatives were prepared from (2-aminophenyl)methanol **9** with 2-(phenylethynyl) benzonitrile **10**.

In an effort to develop an optimal conditions, various reaction parameters were studied for the preparation of **11** *via*

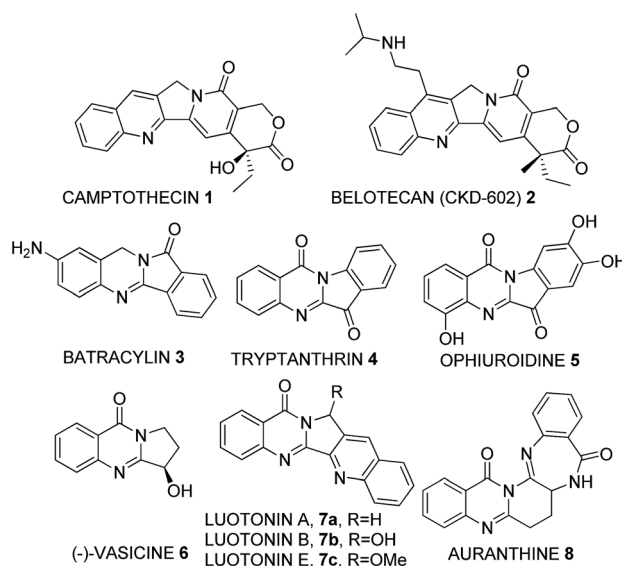
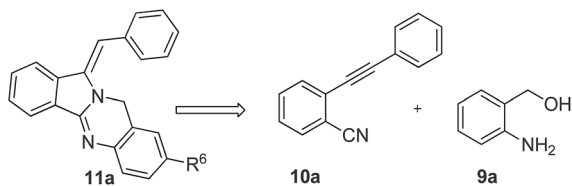


Fig. 1 Examples of natural products containing quinazolinone skeletons.

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Scheme 1 Retrosynthesis of 11a.

Table 1 Screening of various acids^e

Entry	Lewis acid (eq./vol)	Yield ^b (%)
1	BF ₃ ·Et ₂ O ^a (3.0 eq.)	62
2	BF ₃ ·Et ₂ O ^a (1.5 eq.)	49
3	BF ₃ ·Et ₂ O ^a (2.0 eq.)	61
4	BF ₃ ·Et ₂ O ^c (3.0 eq.)	49
5	BF ₃ ·Et ₂ O ^d (3.0 eq.)	26
6	BF ₃ ·2AcOH (3.0 eq.)	
7	Acetic acid (5.0 eq.)	8
8	TFA (5.0 eq.)	16
9	H ₂ SO ₄ (2.0 eq.)	12
10	AlCl ₃ (3.0 eq.)	22
11	AlBr ₃ (3.0 eq.)	18
12	Hg(OAc) ₂ (3.0 eq.)	Traces
13	TiCl ₄ (3.0 eq.)	41

^a 48–50% solution of reagent was used. ^b Isolated yields after column chromatography. ^c Reaction at 45 °C. ^d Reaction with boron trifluoride acetic acid complex at 25 °C. ^e Reaction and conditions: (2-aminophenyl)methanol **9** (1.0 eq.), 2-(phenylethynyl)benzonitrile **10** (1.0 eq.) and BF₃·Et₂O (3.0 eq.) at 70 °C.

Table 2 Screening of solvents^a

Entry	Solvents	Isolated yield (%)
1	DMSO	30
2	DMF	26
3	1,4-Dioxane	62
4	BF ₃ ·Et ₂ O	68
5	Acetonitrile	42
6	THF	15
7	Toluene	20

^a Reaction and conditions: (2-aminophenyl)methanol **9** (1.0 eq.), 2-(phenylethynyl)benzonitrile **10** (1.0 eq.) BF₃·Et₂O (3.0 eq.) at 70 °C.

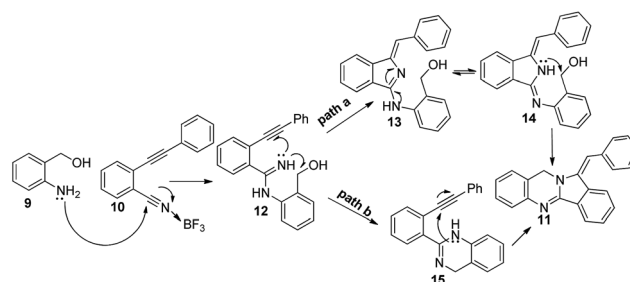
condensation of 2-(phenylethynyl)benzonitrile **10** (1.0 eq.) with (2-aminophenyl)methanol **9** (1.0 eq.) and BF₃·Et₂O (3.0 eq.). The acids have a strong effect on these reactions with respect to yield.

Among all the screened acids, optimum yields were obtained when the reaction was performed in the presence of BF₃·Et₂O (3.0 eq.) (Table 1). Solvents like DMSO, DMF, 1,4-dioxane, THF, acetonitrile and toluene were screened in presence of BF₃·Et₂O. BF₃·Et₂O alone had proven to be the best condition for this reaction instead of use of other solvents (Table 2).

With the optimized reaction conditions in hand, we explored the applicability of our reaction. We employed a variety of

Table 3 Synthesis of various isoindoloquinazolinones derivatives

Entry	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	Product	Yield
1	H	H	H	H	H	H	11a	68
2	H	H	H	CH ₃	H	H	11b	72
3	H	H	H	H	H	Cl	11c	65
4	H	H	CH ₃	CH ₃	H	H	11d	71
5	H	H	CH ₃	H	H	H	11e	69
6	H	H	CH ₃	CH ₃	CH ₃	H	11f	74
7	H	H	H	CH ₃	CH ₃	H	11g	75
8	H	H	H	CH ₃	CH ₃	OCH ₃	11h	69
9	OCH ₃	OCH ₃	H	H	H	H	11i	67
10	OCH ₃	OCH ₃	H	CH ₃	H	H	11j	64



Scheme 2 Proposed reaction mechanism.

substituted alcohols and substituted benzonitriles & the results were summarized in Table 3. Good yields were observed when the reaction was conducted with (2-aminophenyl)propan-2-ol and (2-aminophenyl)ethanol when compared to (2-aminophenyl)methanol due to the stability of the carbocation.

The Scheme 2 represents a plausible mechanism for the three component reaction leading to the compound **11**. The nucleophilic attack of primary amine on nitrile group of **10** yield imidamide intermediate **12**, imidamide can attack on alkyne or alcohol leads to the formation of cyclized intermediate either **13** or **15** which on subsequent cyclization will yield the **11**.

In conclusion, we have established a short and efficient methodology for the synthesis of isoindoloquinazolinone derivatives. The novel synthetic approach involves construction of two new rings *via* sequential C–N bond formation under Lewis acid condition. 4g-scale synthesis of compound **11a** was performed with success. This methodology is operationally simple and amenable for scale-up.

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Notes and references

- 1 *Bioactive Heterocyclic Compound Classes: Pharmaceuticals and Agrochemicals*, ed. C. Lamberth and J. Dinges, Wiley-VCH, New York, 2012.
- 2 (a) S. S. Jew, H. J. Kim, M. G. Kim, E. Y. Roh, Y. S. Cho, J. K. Kim, K. H. Cha, K. K. Lee, H. J. Han, J. Y. Choi and H. Lee, *Bioorg. Med. Chem. Lett.*, 1996, **6**, 845; (b) J. H. Lee, J. M. Lee, J. K. Kim, S. K. Ahn, S. J. Lee, M. Y. Kim, S. S. Jew, J. G. Park and C. Hong, *Arch. Pharmacol. Res.*, 1998, **21**, 581.
- 3 (a) K. Dzierzbicka, P. Trzonkowski, P. L. Sewerynek and A. Mysliwski, *J. Med. Chem.*, 2003, **46**, 978; (b) J. Guillaumel, S. Leonce, A. Pierre, P. Renard, B. Pfeiffer, P. B. Arimondo and C. Monneret, *Eur. J. Med. Chem.*, 2006, **41**, 379; (c) R. Yilin, C. Yun Feng, C. Ting, A. Y. Chen, C. Yu, L. F. Liu and C. C. Cheng, *Pharm. Res.*, 1993, **10**, 918.
- 4 P. Friedländer and N. Roschdestwensky, *Chem. Ber.*, 1915, **48**, 1841.
- 5 N. K. Utkina and S. A. Fedoreev, *Tetrahedron Lett.*, 2007, **48**, 4445–4447.
- 6 (a) V. Ziaee, H. Jalalizadeh, M. Iranshahi and A. Shafiee, *Iran. J. Chem. Chem. Eng.*, 2004, **23**, 33; (b) A. Kamal, V. K. Ramana and M. V. Rao, *J. Org. Chem.*, 2001, **66**, 997.
- 7 (a) T. Harayama, Y. Morikami, Y. Shigeta, H. Abe and Y. Takeuchi, *Synlett*, 2003, 847–848; (b) Z. Ma, Y. Hano, T. Nomura and Y. Chen, *Bioorg. Med. Chem. Lett.*, 2004, **14**, 1193–1196; (c) M. C. Tseng, Y. W. Chu, H. P. Tsai, C. M. Lin, J. Hwang and Y. H. Chu, *Org. Lett.*, 2011, **13**, 920–923; (d) S. B. Mhaske and N. P. Argade, *J. Org. Chem.*, 2004, **69**, 4563–4566; (e) S. P. Chavan and R. Sivappa, *Tetrahedron*, 2004, **60**, 9931–9935; (f) M. B. Wagh, R. Shankar, U. K. S. Kumar and C. H. Gill, *Synlett*, 2011, 84–88; (g) L. Nagarapu, H. K. Gaikwad and R. Bantu, *Synlett*, 2012, **23**, 1775–1778; (h) M.-C. Tseng, P.-Y. Lai, L. Shi, H.-Y. Li, M.-J. Tseng and Y.-H. Chu, *Tetrahydron*, 2014, **70**, 2629; (i) R. Shankar, M. B. Wagh, M. V. Madhubabu, N. Vembu and U. K. S. Kumar, *Synlett*, 2011, 844–848.
- 8 S. E. Yeulet, P. G. Mantle, J. N. Bilton, H. S. Rzepa and R. N. Sheppard, *J. Chem. Soc., Perkin Trans. 1*, 1986, 1891–1894.
- 9 W. W. Li, X. Y. Wang, R. L. Zheng, H. X. Yan, Z. X. Cao, L. Zhong, Z. R. Wang, P. Ji, L. L. Yang, L. J. Wang, Y. Xu, J. J. Liu, J. Yang, C. H. Zhang, S. Ma, S. Feng, Q. Z. Sun, Y. Q. Wei and S. Y. Yang, *J. Med. Chem.*, 2012, **55**, 3852.
- 10 L. A. Mitscher, W. C. Wong, T. De Meulenere, J. Sulko and S. Drake, *Heterocycles*, 1981, **15**, 1017–1019.
- 11 W. R. Bowman, M. R. J. Elsegood, T. Stein and G. W. Weaver, *Org. Biomol. Chem.*, 2007, **5**, 103–113.
- 12 (a) R. K. Raghavendra, M. Ramamohan, A. Raghunadh, M. Suresh Babu, S. Praveen Kumar, D. Kalita, E. Laxminarayana, B. Prasad and M. Pal, *RSC Adv.*, 2015, **5**, 61575; (b) N. V. Murthy, S. P. Nikumbh, S. Praveen Kumar, L. Vaikunta Rao and A. Raghunadh, *Tetrahedron Lett.*, 2015, **56**, 5767–5770.
- 13 J. MinQiang and L. Y. Shu, *ACS Catal.*, 2013, **3**, 622.
- 14 (a) H. Yan, Z. Xinying and F. Xuesen, *Chem. Commun.*, 2014, **50**, 5641; (b) X. Pu, H. Li and T. J. Colacot, *J. Org. Chem.*, 2013, **78**, 568.